

### **experiment soap making saponification pdf**

EXPERIMENT : SOAP MAKING (SAPONIFICATION) In this experiment we prepare soap from animal fat (lard) or vegetable oil. Animal fats and vegetable oils are esters of carboxylic acids; they have a high molecular weight and contain the alcohol, glycerol.

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The objective of this laboratory is to make lye soap via the saponification reaction. Soap making has remained unchanged over the centuries. The ancient Roman tradition called for mixing rain water, potash and animal tallow (rendered form of beef or mutton fat). Making soap was a long and arduous ...

### **12: Making Soap - Saponification (Experiment) - Chemistry**

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Making Soap - Saponification Objectives The objective of this laboratory is to make lye soap via the saponification reaction. Background. Soap making has remained unchanged over the centuries. The ancient Roman tradition called for mixing rain water, potash and animal tallow.

### **Making Soap - Saponification**

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The saponification reaction occurs between an acid and a base, shown in the figure in the procedure. In the reaction you performed, what is the acid and what is the base? 2.

### **Making Soap - Saponification**

Experiment 13 " Preparation of Soap. Soaps are carboxylate salts with very long hydrocarbon chains. Soap can be made from the base hydrolysis of a fat or an oil.

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### **Discussion saponification of soap | Soap | Hydroxide**

The chemicals create reactions: saponification is the primary chemical reaction of soap making. How Saponification Happens Saponification is an exothermic (gives off heat) chemical reaction that occurs when fats or oils (fatty acids) come into contact with lye (a base).

### **Saponification in the Soap Making Process**

The saponification reaction is exothermic in nature, because heat is liberated during the process. The soap

formed remains in suspension form in the mixture. Soap is precipitated as a solid from the suspension by adding common salt to the suspension.

**Saponification-The process of Making Soap (Theory) : Class**

79 Experiment 13 “ Preparation of Soap Soaps are carboxylate salts with very long hydrocarbon chains. Soap can be made from the base hydrolysis of a fat or an oil. This hydrolysis is called saponification, and the reaction has been known for centuries.

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