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An acid is a molecule or ion capable of donating a hydron (proton or hydrogen ion H^+), or, alternatively, capable of forming a covalent bond with an electron pair (a Lewis acid).. The first category of acids is the proton donors or Brønsted acids. In the special case of aqueous solutions, proton donors form the hydronium ion H_3O^+ and are known as Arrhenius acids.

Acid - Wikipedia

Our copper page has over 370 facts that span 107 different quantities. Each entry has a full citation identifying its source. Areas covered include atomic structure, physical properties, atomic interaction, thermodynamics, identification, atomic size, crystal structure, history, abundances, and nomenclature.

Copper | The Periodic Table at KnowledgeDoor

Our chromium page has over 250 facts that span 106 different quantities. Each entry has a full citation identifying its source. Areas covered include atomic structure, physical properties, atomic interaction, thermodynamics, identification, atomic size, crystal structure, history, abundances, and nomenclature.

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La teoria di Brønsted-Lowry è una teoria sulle reazioni acido-base formulata nel 1923 da Johannes Nicolaus Brønsted e Thomas Martin Lowry.. Secondo Brønsted e Lowry, un acido è una specie chimica capace di donare uno ione H^+ ad un'altra specie chimica; similmente, una base è una specie chimica capace di accettare uno ione H^+ da un'altra specie chimica.

Teoria acido-base di Brønsted-Lowry - Wikipedia

Gammateily on ionisoivaa sähkömagneettista säteilyä, joka on peräisin atomiydinten viritystilojen purkautumisesta radioaktiivisessa hajoamisessa tai elektroni-positroniparien annihilaatiosta. Gammateilyä vapautuu suuria määriä gammapurkauksissa.. Gammateilyä syntyy maailmankaikkeudessa monissa rajuissa ilmiöissä kuten kaasun kuumentuessa mustaa aukkoa ympäröivässä ...

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